#### LCS Laboratory Inc.



# Crystalline Silica, Exposure Assessment and Analysis

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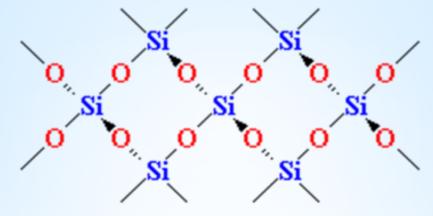
#### PRESENTATION OWERVIEW:

- What is silica?
- Chemistry and structure of silica
- Why should we worry about silica exposure?
- Source of crystalline silica
- Routes of exposure to crystalline silica
- Respirable silica exposure limit
- Sampling procedure
- Introduction into chemical analysis



**Quartz Cristal** 

#### What is Silica?





**Quartz Sand** 

- Word "silica" is synonymous with silicon dioxide (SiO<sub>2</sub>) the chemical compound that is composed of one silicon (Si) atom and two oxygen (O) atoms.
- Silicon and oxygen are the two most abundant elements in the earth's crust
- Silica is one of the most common compound in the earth

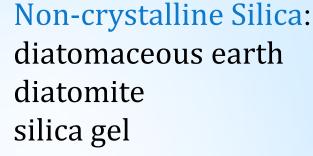
## Types of Silica

Silica exists in two forms:

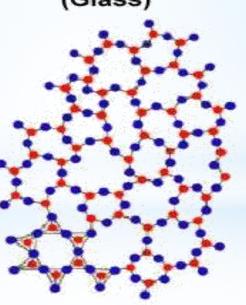
Crystalline silica:

quartz tridymite cristobalite

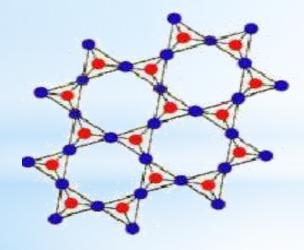
Crystalline SiO<sub>2</sub> (Quartz)



Amorphous SiO<sub>2</sub> (Glass)



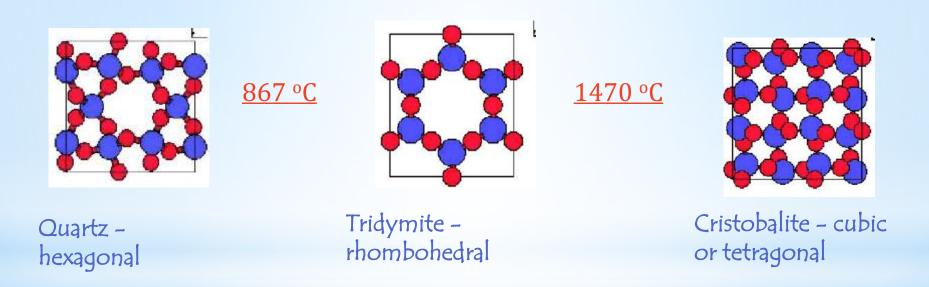
Amorphous silica has the same chemical formula as crystalline silica, but has irregular structure





# The relationships between the polymorphic forms of Crystalline Silica

Silica is polymorphic: it can co-exist in different crystalline forms.



### Silica may cause silicosis

 Silicosis is lung disease caused by breathing in dust that contains silica particles



- Center for Disease Control (USA), MOL (Ontario), OSHA (USA), and all authorities around the Globe, see Airborn Silica Dust as hazardous to our health
- The damage caused by silica inhalation is permanent and irreversible. Symptoms can take years to develop
- Mechanism of silicosis is not well understood. Possibly:
  - Mechanical blockage of lungs
  - Abnormal hardness of Quartz that causes tissue damage
  - Abnormal inertness, that prevents enzymes from it dissolution and excretion

# Industrial sources of crystalline silica:

- ✓ All construction materials that contain natural sand
  - Concrete
  - **Brick**
  - ► Grout and mortar
  - Ceramic and porcelain tiles



- Composite materials
- Aggregate
- ✓ Refractory fibres
- ✓ Refractory bricks
- ✓ Moulding sand





#### People do not recognize danger of silica

- "This is just sand and dust"
  - Crystalline silica is chemically inert
  - >It is naturally occurring
- Word "Silica" does not sound as alarming as "Lead" or "Asbestos"



- It is not common to label silica products as "carcinogenic"
- Silica can appear from "nowhere"

For example:

- ✓ Burning of agricultural waste or products such as rice hulls may cause amorphous silica or Quartz to become Cristobalite
- Rrefractory fibres after years of use at high temperature become Quartz and Cristobalite



#### Respirable Crystalline Silica Exposure Limits:

- **✓ MOL OEL** regulation (O. Reg. 490/09) is:
  - 100 μg/m<sup>3</sup> for Quartz and Tripoli
  - 50 μg/m<sup>3</sup> for Cristobalite
- ✓ OSHA Permissible Exposure Limit (PEL) is 50 µg/m³ of respirable crystalline silica, 8-hour average
- **ACGIH** threshold limit value (**TLV**) is 25 μg/m<sup>3</sup>

Silica is always regulated as <u>Respirable Fraction</u> and must be collected as Respirable Dust (<10 microns)

### Sampling and Monitoring

- Several standard methods are available: NIOSH (USA), OSHA (USA), and MDHS (UK)
  - All methods use personal sampling pumps
  - All methods use cyclones to separate Respirable Fraction
  - Difference between the methods is in type of filter used, sample preparation and analysis





#### LCS Laboratory recommends NIOSH 7602

- Samples are collected on 3-piece 37 mm diameter cassette
- Dust is filtrated on PVC membrane with 5 micron pore size
- Respirable fraction of dust is separated with a cyclone
- Air is collected using laboratory calibrated, portable, sampling pump
- Sample can be collected anywhere from 2 to 10 hours
- Sample can be collected as "Personal" or "Area" samples
- Samples should be returned to LCS Lab for analysis.



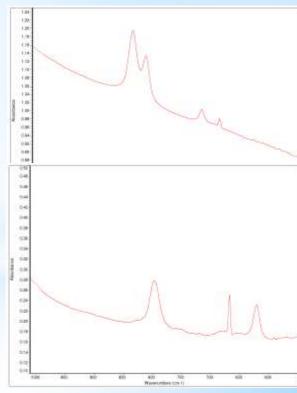




#### How we analyze your sample

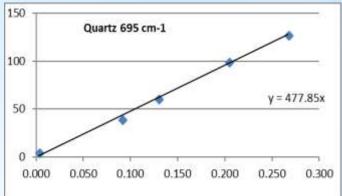
- We extract filter with accumulated dust from the cassette
- We wash the dust to remove alkaline impurities, which may react with quartz and turn it into silicates (glass)
- ➤ We ash the filter at 500°C to remove the filter itself and any organic impurities
- We mix the residual dust with potassium bromide, and make 13 mm disks
- We calibrate our instrument with set of Quartz and Cristobalite samples of known concentration

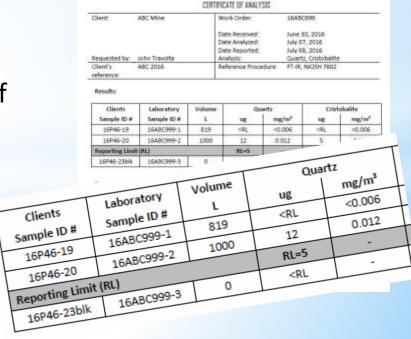




#### How we calculate and report the results

- We measure signal at different wavelengths
- We recalculate signal strength into Quartz and Cristobalite concentration in micrograms per sample
- We calculate volume of air that you collected in liters
- We calculate and report you our findings in form of formal certificate of analysis
- The results can be compared against MOL OEL, to see if you are in compliance





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